

(2)

2. (a) Explain methods of the simplification of complex NMR spectra. 10
 (b) Explain COSY, NOESY, APT and DEPT techniques. 10
3. (a) Discuss the applications of resonance Raman Spectroscopy. 6
 (b) Explain the following : 14
 (i) Hyperfine coupling
 (ii) Significance of g-tensors
4. (a) Explain the principle and instrumentation of mass spectroscopy. 10
 (b) Explain the factors affecting fragmentation. 5
 (c) What are metastable peaks ? 5
5. (a) How can you calculate λ_{\max} value of conjugated dienes and conjugated carbonyl compounds by Fieser-Woodward rule ? 8
 (b) Explain the various electronic transitions which take place between 185-800 nm. 8
 (c) Explain Beer-Lambert law. 4
6. (a) Write note on circular dichroism and its applications. 8

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| (b) Explain following : | 12 |
| (i) Nuclear magnetic double resonance | |
| (ii) Fourier transform technique | |
| 7. (a) Discuss types of photochemical reaction.
Derive equation for rate constant of any
one photochemical reaction. | 8 |
| (b) Explain following photochemical
reactions of aromatic compounds : | 12 |
| (i) Isomerisation | |
| (ii) Addition reaction | |
| (iii) Substitution reaction | |
| 8. (a) Explain following photochemical
reactions of carbonyl compounds : | 12 |
| (i) Norrish type I reaction | |
| (ii) Norrish type II reaction | |
| (iii) Oxetane formation | |
| (b) Discuss the photochemistry of vision and
formation of photochemical smog. | 8 |
| 9. (a) How will you differentiate among metal,
insulators and somiconductors ? | 4 |
| (b) Explain p-n junction. | 10 |
| (c) Write a note on point defects. | 6 |

(4)

10. Explain the following : 20

- (a) Optical properties of solids
 - (b) Organic charge transfer complex
 - (c) Thermodynamics of Frankel defect
 - (d) Colour centers
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