



**PB-164**

M.Sc. Chemistry (1st Semester)  
Examination, Dec., 2018

**Paper - III**

**Physical Chemistry - I**

*Time : Three Hours] [Maximum Marks : 80*  
*[Minimum Pass Marks : 29*

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**Note :** Answer from **both** the Sections as directed. The figures in the right-hand margin indicate marks.

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**SECTION-A**

1. Write the answer of the following questions :  $1 \times 10$ 
  - (a) What is the difference between classical mechanics and quantum mechanics.
  - (b) Define the spin antisymmetry.
  - (c) Write the term symbol  $^3F_2$  by spin multiplicity or spectroscopic state.
  - (d) Draw energy diagram of kinetics of enzymatic reaction.
  - (e) Write the full form of RRKM.

( 2 )

- (f) Discuss the various chain structure of polymers.
- (g) Explain the probable density of postulates of quantum mechanics.
- (h) What is the equation of Laplacian operator in cartesian coordinates ?
- (i) What is conjugated system by molecular orbital theory ?
- (j) Describe the irreversible and reverse micelles.

2. Answer the following questions : 2×5

- (a) Explain the perturbation theory in first order degenerates.
- (b) What is Slater-Condon parameters ?
- (c) Explain the dynamic of barrier less chemical reaction in solution with examples.
- (d) Describe the electrokinetic phenomenon of surface film.
- (e) Give the mechanism of polymerization.

### SECTION-B

Answer all questions : 12×5

3. (a) Explain the harmonic oscillator in hydrogen atom in a box.

( 3 )

- (b) Discuss the eigen function for angular momentum.

**OR**

- (a) Give the importance, introduction of postulate of quantum mechanics.  
(b) What is linear variation principle ?

4. Write short notes on the following :

- (a) Russell-Saunders Term  
(b) Bond order and charge density calculations

**OR**

Explain in brief the term separation energy for  $d^n$  configurations.

5. Discuss activated complex theory and steady state kinetics.

**OR**

Answer the following :

- (a) Pyrolysis of  $\text{CH}_3\text{CHO}$   
(b) Hydrogen-Chlorine reaction  
6. Describe the Gibbs absorption isotherm and Calvin equation.

**OR**

( 4 )

- (a) Discuss the factors affecting the CMC of surfactant.
  - (b) Thermodynamics of micellization phase separation.
7. Give some important methods for the determination of molecular mass.

**OR**

Write short notes on the following :

- (a) Fire resistant
  - (b) Kinetic of polymerization
  - (c) Types of polymers
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